

SYNTHESIS OF 2,3-DISUBSTITUTED  
HEXAHYDRO-BENZO/a/QUINOLIZINES AND  
OCTAHYDRO-INDOLO/2,3-a/QUINOLIZINES

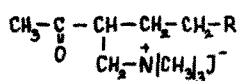
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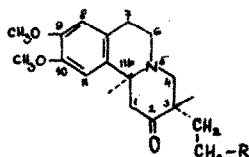
Recently we have published the synthesis of different benzo/a/quinolizine and indolo/2,3-a/quinolizine alkaloids /2,3/. We now wish to report the synthesis of several intermediates, which seem to be useful in the preparation of alloberban and alloyohimban derivatives.

Applying the method of Openshaw and Whittaker /4/, the salt Ia /3/ was reacted with 6,7-dimethoxy-3,4-dihydro-isoquinoline yielding 78 % ester IIa /m.p. 130° /from ethanol/;  $\nu_{\text{max}}^{\text{KBr}}$  1735 /COOCH<sub>3</sub>/, 1705 /CO/; found C: 66,03; H: 7,34; N: 4,07; C<sub>19</sub>H<sub>25</sub>NO<sub>5</sub>; requires C: 65,79; H: 7,25; N: 4,03./.

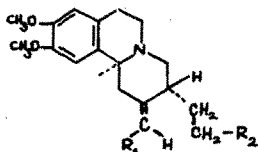
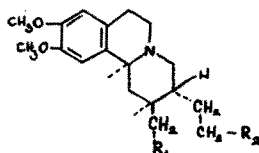
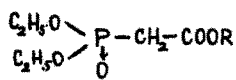
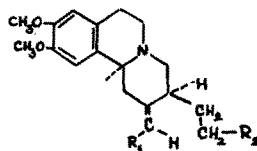
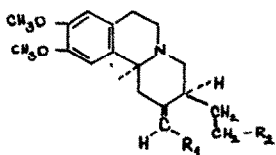
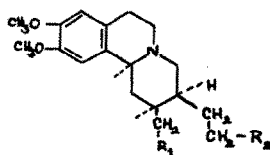
The Wittig reaction of IIa with ethoxycarbonylmethylene-triphenylphosphorane at 150° furnished the ester IIIa in 83 % yield /cf. 5,6/ /m.p. 128° /from ethanol/;  $\nu_{\text{max}}^{\text{KBr}}$  1738 /COOCH<sub>3</sub>/ 1710 /COOC<sub>2</sub>H<sub>5</sub> conj./; 1642 /C=C/; found: C: 66,05; H: 7,47; N: 3,67; C<sub>23</sub>H<sub>31</sub>NO<sub>6</sub>; requires: C: 66,16;

Ia: R = COOCH<sub>3</sub>

b: R = CN

IIa: R = COOCH<sub>3</sub>

b: R = CN

IIIa: R<sub>1</sub> = COOC<sub>2</sub>H<sub>5</sub> R<sub>2</sub> = COOCH<sub>3</sub>b: R<sub>1</sub> = COOC<sub>2</sub>H<sub>5</sub> R<sub>2</sub> = CNIVa: R<sub>1</sub> = COOC<sub>2</sub>H<sub>5</sub> R<sub>2</sub> = COOCH<sub>3</sub>b: R<sub>1</sub> = R<sub>2</sub> = COOCH<sub>3</sub>c: R<sub>1</sub> = COOC<sub>2</sub>H<sub>5</sub> R<sub>2</sub> = CNVa: R = C<sub>2</sub>H<sub>5</sub>b: R = CH<sub>3</sub>VIa: R<sub>1</sub> = COOC<sub>2</sub>H<sub>5</sub> R<sub>2</sub> = COOCH<sub>3</sub>b: R<sub>1</sub> = COOC<sub>2</sub>H<sub>5</sub> R<sub>2</sub> = CNVIIa: R<sub>1</sub> = COOC<sub>2</sub>H<sub>5</sub> R<sub>2</sub> = COOCH<sub>3</sub>b: R<sub>1</sub> = COOC<sub>2</sub>H<sub>5</sub> R<sub>2</sub> = CNVIIIa: R<sub>1</sub> = COOC<sub>2</sub>H<sub>5</sub> R<sub>2</sub> = COOCH<sub>3</sub>b: R<sub>1</sub> = COOC<sub>2</sub>H<sub>5</sub> R<sub>2</sub> = CNc: R<sub>1</sub> = R<sub>2</sub> = COOCH<sub>3</sub>

H: 7,48; N: 3,35/. Catalytic /Pd/C/ hydrogenation of IIIa gave the ester IVa /yield: 57 %/ /m.p. 81° /light petroleum/;  $\nu_{\text{max}}^{\text{KBr}}$  1740 and 1730 /ester CO/; found: C: 66,02; H: 7,89; N: 3,64;  $\text{C}_{23}\text{H}_{33}\text{NO}_6$ ; requires: C: 65,85; H: 7,93; N: 3,34/, /IVa.HCl: m.p. 182° /from methanol-ether/;  $\nu_{\text{max}}^{\text{KBr}}$  1735 and 1715 /ester CO/; found: C: 60,26; H: 7,73; N: 3,21;  $\text{C}_{23}\text{H}_{33}\text{NO}_6\cdot\text{HCl}$ ; requires: C: 60,58; H: 7,51; N: 3,07./

The ester IVa was transformed into the dimethylester IVb /m.p. 96-97° /from methanol-water/,  $\nu_{\text{max}}^{\text{KBr}}$  1735 and 1730 / $\text{COOCH}_3$ / found: C: 65,18; H: 7,63; N: 3,45;  $\text{C}_{22}\text{H}_{31}\text{NO}_6$ ; requires: C: 65,16; H: 7,71; N: 3,45./ by boiling it in methanol in the presence of sodium methoxide.

When the condensation reaction of IIa was carried out with the reactant Va in DMF, in the presence of excess K-tert.-butoxide, together with IIIa compounds VIa /m.p. 104° /from ethanol/;  $\nu_{\text{max}}^{\text{KBr}}$  1730 / $\text{COOCH}_3$ /, 1710 / $\text{COOC}_2\text{H}_5$  conj./, 1640 /C=C/; found: C: 66,23; H: 7,32; N: 3,33;  $\text{C}_{23}\text{H}_{31}\text{NO}_6$ ; requires: C: 66,16; H: 7,48; N: 3,35/, and VIIa /m.p. 101° /from ethanol/;  $\nu_{\text{max}}^{\text{KBr}}$  1740 / $\text{COOCH}_3$ /; 1715 / $\text{COOC}_2\text{H}_5$  conj./, 1640 /C=C/; found: C: 66,28; H: 7,42; N: 3,57;  $\text{C}_{23}\text{H}_{31}\text{NO}_6$ ; requires: C: 66,16; H: 7,48; N: 3,35/, were isolated in about 15 % yield.

Catalytic /Pd/C/ hydrogenation of VIa or VIIa in dioxane solution gave the same ester VIIIa /VIIIa.HCl m.p. 179-180° /from methanol-ether/;  $\nu_{\text{max}}^{\text{KBr}}$  1735 and 1730 /ester CO/; found: C: 60,75; H: 7,58; N: 3,24;  $\text{C}_{23}\text{H}_{33}\text{NO}_6\cdot\text{HCl}$ ;

requires: C: 60,58; H: 7,51; N: 3,07/, as the only product; the hydrochloride of which depressed the m.p. of hydrochloride of IVa.

The salt Ib /m.p. 143-147° /from methanol-ether/;  
 $\nu_{\text{max}}^{\text{KBr}}$  2245 /CN/; 1715 /CO/; found C: 38,83; H: 6,12;  
 N: 8,46; I: 40,66;  $\text{C}_{10}\text{H}_{19}\text{N}_2\text{OI}$ ; requires: C: 38,72; H: 6,17;  
 N: 9,03; I: 40,91/, was prepared from 5-oxocaproic-acid-  
 -nitril /7/ by Mannich reaction /formaldehyd + dimethylami-  
 ne hydrochloride/ and treatment of the obtained base with  
 methyl iodide, in 36 % overall yield.

Reaction of 6,7-dimethoxy-3,4-dihydro-izoquinoline ba-  
 se with the salt Ib in boiling methanol furnished the keto-  
 ne IIb /m.p. 130-131° /from methanol/,  $\nu_{\text{max}}^{\text{KBr}}$  2250 /CN/; 1710  
 /CO/; found: C: 68,71; H: 6,92; N: 8,99;  $\text{C}_{18}\text{H}_{22}\text{N}_2\text{O}_3$ ; re-  
 quires: C: 68,76; H: 7,05; N: 8,91/, in 64 % yield. The  
 latter was reacted with Va in DMF, in the presence of ex-  
 cess K.tert.-butoxide. Fractional crystallisation of the  
 obtained product from ethanol gave isomers VIb /m.p. 140-  
 -141° /from ethanol/,  $\nu_{\text{max}}^{\text{KBr}}$  2245 /CN/; 1715 /COOC<sub>2</sub>H<sub>5</sub> conj./,  
 1645 /C=C/; NMR: /in CDCl<sub>3</sub>/  $\delta$ : 6,81 /C-11 proton/, 6,59  
 /C-8 proton/, 5,86 /olefinic proton/, 4,50 /C-11b proton/,  
 4,20 /ester methylene protons/, quartet, J=7 c.p.s./, 3,86  
 /methoxyl protons/, 1,31 /ester methyl protons, triplet,  
 J=7 c.p.s./, found: C: 68,61; H: 7,41; N: 7,33;  $\text{C}_{22}\text{H}_{28}\text{N}_2\text{O}_4$ ;  
 requires: C: 68,72; H: 7,34; N: 7,29/, and VIIb /m.p. 131-  
 -132°,  $\nu_{\text{max}}^{\text{KBr}}$  2245 /CN/; 1715 /COOC<sub>2</sub>H<sub>5</sub> conj./, 1640 /C=C/;  
 NMR: /in CDCl<sub>3</sub>/  $\delta$ : 6,61 /aromatic protons/, 5,94 /olefinic

proton/ 4,18 /ester methylene protons, quartet,  $J=7$  c.p.s./, 3,85 /methoxy protons/, 1,29 /ester methyl protons, triplet,  $J=7$  c.p.s./, found: C: 68,48; H: 7,42; N: 7,21;  $C_{22}H_{28}N_2O_4$ ; requires: C: 68,82; H: 7,34; N: 7,28/ in 64 % yield together with IIIb obtained in 20 % yield /m.p. 109-110°;  $\nu_{\text{max}}^{\text{KBr}}$  2248 /CN/; 1710 /COOC<sub>2</sub>H<sub>5</sub> conj./, 1650 /C=C/; NMR: /in CDCl<sub>3</sub>/  $\delta$ : 6,84 /C-11 proton/, 6,61 /C-8 proton/, 5,61 /olefinic proton/, 4,42 /C-11b proton/, 4,22 /ester methylene protons, quartet  $J=c.p.s.$ /, 3,89 and 3,84 /methoxyl protons/, 1,31 /ester methyl protons, triplet,  $J=7$  c.p.s./, found: C: 68,78; H: 7,29; N: 7,27;  $C_{22}H_{28}N_2O_4$ ; requires: C: 68,72; H: 7,34; N: 7,29%. The compound IIIb was identical with the product isolated from the Wittig reaction of IIb with ethoxycarbonylmethylene-triphenylphosphorane /6/.

The stereochemistry of compounds IIIb, VIb, and VIIb was substantiated by n.m.r. data. The aromatic proton at C-11 of ester IIIb / $\delta=6,84$ / and VIb / $\delta=6,81$ / resonates at higher field than that of VIIb / $\delta=6,61$ / due to the deshielding effect of ester-carbonyl group, while the signal of the other aromatic proton /at C-8/ remains unchanged. Furthermore, the ester VIIb gives one singlet for the protons of the methoxyl group in positions 9 and 10 / $\delta=-3,82$ /, while they show separate /3 c.p.s./ peaks in compounds IIIb and VIb due to the same effect.

Catalytic /Pd/C/ hydrogenation of compounds VIb and VIIb yielded the nitrile ester VIIIb /m.p. 110-111° /from ethanol-water/,  $\nu_{\text{max}}^{\text{KBr}}$  2245 /CN/; 1710 /COOC<sub>2</sub>H<sub>5</sub>/; found:

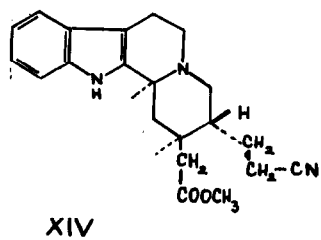
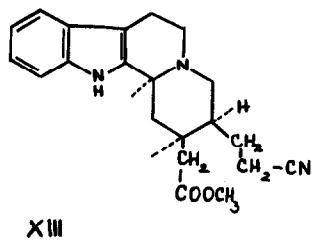
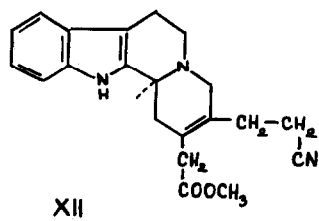
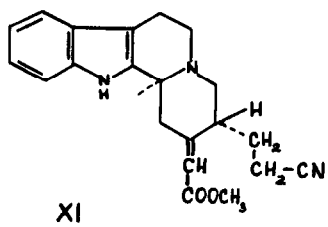
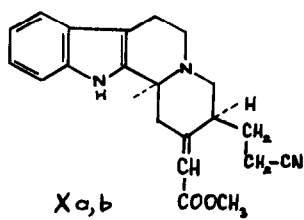
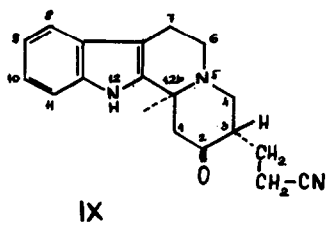
C: 68,43; H: 8,05; N: 7,47;  $C_{22}H_{30}N_2O_4$ ; requires: C: 68,37  
 H: 7,82; N: 7,25;/ similar treatment of IIIb lead to IVc  
 /m.p. 109° /from ethanol-water/,  $\nu_{\text{max}}^{\text{KBr}}$  2245 /CN/; 1720  
 /COOC<sub>2</sub>H<sub>5</sub>/; found: C: 68,33; H: 7,59; N: 7,49;  $C_{22}H_{30}N_2O_4$ ;  
 requires: C: 68,37; H: 7,82; N: 7,25/.

On boiling the nitrilester VIIIb in a methanol-sulfu-  
 ric acid mixture the dimethylester VIIIC /m.p. 63-64° /from  
 light petrolether/,  $\nu_{\text{max}}^{\text{KBr}}$  1740 and 1730 /COOCH<sub>3</sub>/, found:  
 C: 64,90; H: 7,75; N: 3,57;  $C_{22}H_{31}NO_6$ ; requires: C: 65,16;  
 H: 7,71; N: 3,45/ was obtained in 71 % yield.

Similar reaction sequences were carried out with the  
 appropriate indol derivatives.

On boiling 3,4-dihydro- $\beta$ -carboline /8/ with Ib in  
 methanol, the compound IX /m.p. 214-215° /from dioxane-wa-  
 ter/,  $\nu_{\text{max}}^{\text{KBr}}$  3370 /NH/; 1705 /CO/; found: C: 73,50; H: 6,50;  
 N: 14,21;  $C_{18}H_{19}N_2O$ ; requires: C: 73,69; H: 6,53; N: 14,33/  
 was obtained in 57 % yield.

The reaction of ketone IX with Vb in DMF, in the pre-  
 sence of sodium-methoxide at 0° yield a mixture of four  
 isomers in 80 % yield. The components of the mixture could  
 be separated on the basis of their differential solubility  
 in methanol giving to the olefinic-bond cis-trans isomers  
 Xa /m.p. 175-176° /from methanol/,  $\nu_{\text{max}}^{\text{KBr}}$  3370 /NH/; 2245  
 /CN/; 1710 /COOCH<sub>3</sub>; conj./, 1640 /C=C/; found: C: 71,91;  
 H: 6,75; N: 11,67;  $C_{21}H_{23}N_3O_2$ ; requires: C: 72,18; H: 6,63;  
 N: 12,03; in 6 % Xb /m.p. 98-125° /from methanol-water/,  
 $\nu_{\text{max}}^{\text{KBr}}$  3360 /NH/; 2240 /CN/; 1710 /COOCH<sub>3</sub> conj./; 1655 /C=C/;



found: C: 65,78; H: 6,91; N: 11,40;  $C_{21}H_{23}N_3O_2 \cdot 1/5 H_2O$ ;  
 requires: C: 67,03; H: 6,96; N: 11,08; / in 23 %, and XI  
 /m.p.  $101^\circ$  /from methanol/,  $\nu_{\text{max}}^{\text{KBr}}$  3370 /NH/; 2245 /CN/;  
 1710 /COOCH<sub>3</sub> conj./; 1645 /C=C/; found: C: 68,98; H: 6,96;  
 N: 11,05;  $C_{21}H_{23}N_3O_2 \cdot CH_3OH$ ; requires: C: 69,27; H: 7,13;  
 N: 11,02/ in 16 % yield, respectively.

In addition the compound XII, which contains the double bond in an endocyclic position, /9/ was also isolated.  
 /m.p.  $194-196^\circ$  /from dioxane-water/;  $\nu_{\text{max}}^{\text{KBr}}$  3360 /NH/; 2245  
 /CN/; 1720 /COOCH<sub>3</sub>/; found: C: 72,05; H: 6,35; N: 12,01;  
 $C_{21}H_{23}N_3O_2$ ; requires: C: 72,18; H: 6,63; N: 12,03/.

Catalytic /Pd/C/ reduction of Xa and Xb in dioxane solution afforded the same product XIII /m.p.  $145-146^\circ$  /from methanol-water/;  $\nu_{\text{max}}^{\text{KBr}}$  3390 /NH/, 2245 /CN/; 1710 /COOCH<sub>3</sub>/; found: C: 71,64; H: 7,00; N: 11,85;  $C_{21}H_{25}N_3O_2$ ; requires: C: 71,77; H: 7,17; N: 11,96/.

The hydrogenation of XI in methanol in the presence of Pd/C gave the ester XIV /m.p.  $177-178^\circ$  /from methanol-water/;  $\nu_{\text{max}}^{\text{KBr}}$  3335 /NH/; 2250 /CN/; 1730 /COOCH<sub>3</sub>/; found: C: 72,02; H: 7,28; N: 11,82;  $C_{21}H_{25}N_3O_2$ ; requires: C: 71,77; H: 7,17; N: 11,96/.

Further investigations with these compounds are in progress.



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